

## HYDROGEN ATOM: RADIAL FUNCTIONS FOR LARGE L

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Reference: Griffiths, David J. (2005), Introduction to Quantum Mechanics, 2nd Edition; Pearson Education - Problem 4.46.

In solving the radial equation for the hydrogen atom, we arrived at the solution

$$(1) \quad u(\rho) = \rho^{l+1} e^{-\rho} v(\rho)$$

$$(2) \quad u(r) \equiv rR(r)$$

$$(3) \quad \rho = \kappa r$$

$$(4) \quad \rho_0 = \frac{me^2}{2\pi\epsilon_0\hbar^2\kappa}$$

$$(5) \quad \kappa = \frac{\sqrt{-2mE}}{\hbar}$$

with  $v(\rho)$  given by a series

$$(6) \quad v(\rho) = \sum_{j=0}^{\infty} c_j \rho^j$$

The coefficients turned out to satisfy the recursion formula

$$(7) \quad c_{j+1} = \frac{2(j+l+1) - 2n}{(j+1)(j+2(l+1))} c_j$$

For  $l = n - 1$ , the recursion formula becomes

$$(8) \quad c_{j+1} = \frac{2j}{(j+1)(j+2n)} c_j$$

where we need to specify  $c_0$  to get the series started. Working back to the radial function gives us

$$(9) \quad R_{n(n-1)}(r) = \frac{c_0}{(an)^n} r^{n-1} e^{-r/an}$$

We can combine the constant out front into a single constant  $N_n = c_0/(an)^n$ .

To normalize, we evaluate the integral (using software)

$$(10) \quad \int_0^\infty R_{n(n-1)}^2(r) dr = \frac{c_0^2}{(an)^{2n}} \int_0^\infty r^{2(n-1)} e^{-2r/an} r^2 dr$$

$$(11) \quad = \frac{na(2n)!}{2^{2n+1}} c_0^2$$

$$(12) \quad = 1$$

$$(13) \quad c_0 = 2^n \sqrt{\frac{2}{an(2n)!}}$$

$$(14) \quad N_n = 2^n \sqrt{\frac{2}{(an)^{2n+1}(2n)!}}$$

Expectation values are calculated in the usual way:

$$(15) \quad \langle r \rangle = \frac{2^{2n+1}}{(an)^{2n+1}(2n)!} \int_0^\infty r^{2(n-1)+3} e^{-2r/an} dr$$

$$(16) \quad = \frac{(2n+1)na}{2}$$

$$(17) \quad \langle r^2 \rangle = \frac{2^{2n+1}}{(an)^{2n+1}(2n)!} \int_0^\infty r^{2(n-1)+4} e^{-2r/an} dr$$

$$(18) \quad = \frac{(n+1)(2n+1)n^2 a^2}{2}$$

$$(19) \quad \sigma_r = \sqrt{\langle r^2 \rangle - \langle r \rangle^2}$$

$$(20) \quad = \frac{na\sqrt{2n+1}}{2}$$

$$(21) \quad = \frac{\langle r \rangle}{\sqrt{2n+1}}$$

A few radial functions  $R_{n(n-1)}(r)$  are shown. The domain of  $r$  is up to  $200a$  in each case.





