SPONTANEOUS EMISSION RATES FOR THE HYDROGEN **ATOM**

Link to: physicspages home page.

To leave a comment or report an error, please use the auxiliary blog.

References: Griffiths, David J. (2005), Introduction to Quantum Mechanics, 2nd Edition; Pearson Education - Problem 9.10, 9.11.

A collection of atoms an excited state will decay over time into the ground state by spontaneous emission of radiation. If the rate of decay is A (this is the rate at which a single atom decays, defined as the derivative of the transition probability with respect to time), then over a time dt a fraction A dt of the number N_b of atoms still in the excited state will decay (on average). That is

$$dN_b = -AN_b dt (1)$$

which is the ODE for simple exponential decay, so

$$N_b(t) = N_b(0) e^{-At} \tag{2}$$

The time constant is

$$\tau = \frac{1}{A} \tag{3}$$

and is the time required for the original population to be reduced to $1/e \approx$ 0.368 of its size at t = 0. An often-quoted, related quantity is the half-life $t_{1/2}$ which is the time required for the population to be reduced to half its original size. That is

$$e^{-At_{1/2}} = 0.5 (4)$$

$$e^{-At_{1/2}} = 0.5$$
 (4)
 $t_{1/2} = -\frac{\ln 0.5}{A}$ (5)
 $= \frac{\ln 2}{A}$ (6)

$$= \frac{\ln 2}{A} \tag{6}$$

$$\approx 0.693\tau$$
 (7)

As an example, we can calculate the lifetimes of the four n = 2 states of the hydrogen atom. The transition rate is

$$A = \frac{\omega_0^3 |\mathbf{p}|^2}{3\epsilon_0 \pi \hbar c^3} \tag{8}$$

The main calculation is the evaluation of the matrix elements

$$\mathbf{p} = q \left\langle \psi_b \left| \mathbf{r} \right| \psi_a \right\rangle \tag{9}$$

where q is the electron charge and ψ_i is one of the hydrogen wave functions, given by (with $a = 4\pi\epsilon_0\hbar^2/mq^2$ as the Bohr radius)

$$\psi_{nlm} = \sqrt{\left(\frac{2}{na}\right)^3 \frac{(n-l-1)!}{2n\left[(n+l)!\right]^3}} e^{-r/na} \left(\frac{2r}{na}\right)^l L_{n-l-1}^{2l+1} \left(\frac{2r}{na}\right) Y_l^m(\theta,\phi)$$
(10)

where L is an associated Laguerre polynomial and Y is a spherical harmonic. Doing these integrals by hand gets quite tedious and error-prone, so it's easier to use Maple to do them. The Maple code for evaluating the Laguerre polynomial is

$$simplify\left(LaguerreL\left(n-l-1,2*l+1,\frac{2*r}{(n*a)}\right),'LaguerreL'\right)$$
(11)

The spherical harmonic can be evaluated using

$$simplify (convert(SphericalY(l, m, t, p), Legendre P), 'Legendre P')$$
(12)

The t stands for θ and the p stands for ϕ .

With these definitions, we can get expressions for ψ_{200} , ψ_{21-1} , ψ_{210} , ψ_{211} and ψ_{100} and use the rectangular to spherical conversion formulas

$$x = r\sin\theta\cos\phi \tag{13}$$

$$y = r \sin \theta \sin \phi \tag{14}$$

$$z = r\cos\theta \tag{15}$$

With $\psi_a = \psi_{100}$ and ψ_b set to each of the four n=2 wave functions in turn, we have a total of 12 integrals to work out to evaluate 9 in all cases. We can do all these in Maple, but here's the explicit formula for one of the integrals so you can see what they look like.

$$\langle \psi_{210} | z | \psi_{100} \rangle = \int_0^\infty dr \int_0^\pi d\theta \int_0^{2\pi} d\phi r^2 \sin\theta \left[r \cos\theta \psi_{210}^* \psi_{100} \right]$$
(16)
$$= \int_0^\infty dr \int_0^\pi d\theta \int_0^{2\pi} d\phi r^2 \sin\theta \left(r \cos\theta \right) \left[\frac{\sqrt{2}}{8a^{5/2} \sqrt{\pi}} r e^{-r/2a} \cos\theta \right] \left[\frac{e^{-r/a}}{a^{3/2} \sqrt{\pi}} \right]$$
(17)
$$= \frac{128\sqrt{2}}{243} a$$
(18)

The other non-zero matrix elements are

$$\langle \psi_{211} | x | \psi_{100} \rangle = \frac{128}{243} ai$$
 (19)

$$\langle \psi_{211} | y | \psi_{100} \rangle = \frac{128}{243} a$$
 (20)

$$\langle \psi_{21-1} | x | \psi_{100} \rangle = \frac{128}{243} ai$$
 (21)

$$\langle \psi_{21-1} | y | \psi_{100} \rangle = -\frac{128}{243} a$$
 (22)

From these we can evaluate 9 for each state.

$$\mathfrak{p}_{200} = 0 \tag{23}$$

$$\mathfrak{p}_{210} = \frac{128\sqrt{2}}{243}qa\hat{\mathbf{z}} \tag{24}$$

$$\mathbf{p}_{211} = \frac{128}{243} qai\hat{\mathbf{x}} + \frac{128}{243} qa\hat{\mathbf{y}}$$
 (25)

$$\mathbf{p}_{21-1} = \frac{128}{243} qai\hat{\mathbf{x}} - \frac{128}{243} qa\hat{\mathbf{y}}$$
 (26)

From this we get $|\mathbf{p}|^2 = |\mathbf{p}_x|^2 + |\mathbf{p}_y|^2 + |\mathbf{p}_z|^2$.

$$\left|\mathfrak{p}_{200}\right|^2 = 0 \tag{27}$$

$$|\mathbf{p}_{210}|^2 = 2\left(\frac{128}{243}aq\right)^2 \tag{28}$$

$$|\mathbf{p}_{211}|^2 = 2\left(\frac{128}{243}aq\right)^2 \tag{29}$$

$$|\mathbf{p}_{21-1}|^2 = 2\left(\frac{128}{243}aq\right)^2 \tag{30}$$

To evaluate the transition rate 8 we need ω_0 which is the energy of the emitted photon, which we can get from the difference in energy between the n=2 and n=1 levels in the Bohr formula

$$E_n = -\frac{1}{n^2} \frac{mq^4}{2\hbar^2 (4\pi\epsilon_0)^2}$$
 (31)

We have

$$\omega_0 = \frac{E_2 - E_1}{\hbar} \tag{32}$$

$$= \frac{3mq^4}{8\hbar^3(4\pi\epsilon_0)^2} \tag{33}$$

$$= 1.55 \times 10^{16} \,\mathrm{s}^{-1} \tag{34}$$

Plugging these results into 8 we get for all states except ψ_{200} :

$$\tau = \frac{1}{A} = 1.595 \times 10^{-9} \,\mathrm{s} \tag{35}$$

For ψ_{200} , the lifetime is $1/A = \infty$ so this state is stable.

PINGBACKS

Pingback: Spontaneous emission from n=3 to n=1 in hydrogen