

## TEMPERATURE DEPENDENCE OF PRESSURE IN PHI-4 THEORY

Link to: physicspages home page.

To leave a comment or report an error, please use the auxiliary blog.

Reference: Tom Lancaster and Stephen J. Blundell, *Quantum Field Theory for the Gifted Amateur*, (Oxford University Press, 2014), Problem 25.3.

Post date: 15 Sep 2019.

In L&B's example 25.4, they derive expressions for the partition function, energy and pressure of a statistical system in free scalar field theory. Their derivation skims over a few points that are worth clarifying in order that we can do their problem 25.3.

For free field theory (not  $\phi^4$  theory yet!), the partition function is given by their equation 25.35:

$$\ln Z = \mathcal{V} \int \frac{d^3p}{(2\pi)^3} \left[ -\frac{1}{2}\beta E_{\mathbf{p}} - \ln \left( 1 - e^{-\beta E_{\mathbf{p}}} \right) \right] \quad (1)$$

where  $\beta = \frac{1}{k_B T}$ , with  $k_B$  the Boltzmann constant and  $T$  the temperature. From here, they derive the Helmholtz energy  $F$ . Although I've covered the Helmholtz energy in thermodynamics, I haven't yet done any posts on statistical mechanics, so the formula in which  $F$  is derived from the partition function that L&B assume the reader knows at this point needs to be spelled out. It can be found in any textbook on statistical mechanics, and is

$$F = -\frac{1}{\beta} \ln Z \quad (2)$$

This gives us the formula in Example 25.4:

$$F = \frac{\mathcal{V}}{\beta} \int \frac{d^3p}{(2\pi)^3} \left[ \frac{1}{2}\beta E_{\mathbf{p}} + \ln \left( 1 - e^{-\beta E_{\mathbf{p}}} \right) \right] \quad (3)$$

Next, they state the ground state energy, which they derive from the formula

$$E_0 = -\frac{\partial}{\partial \beta} \ln Z \quad (4)$$

and which they just state as

$$E_0 = \frac{\mathcal{V}}{2} \int \frac{d^3p}{(2\pi)^3} E_{\mathbf{p}} \quad (5)$$

This requires a bit of explanation. Starting with 1, we take the derivative with respect to  $\beta$  to get

$$-\frac{\partial}{\partial\beta} \ln Z = \mathcal{V} \int \frac{d^3p}{(2\pi)^3} \left[ \frac{1}{2} E_{\mathbf{p}} + \frac{E_{\mathbf{p}} e^{-\beta E_{\mathbf{p}}}}{1 - e^{-\beta E_{\mathbf{p}}}} \right] \quad (6)$$

$$= \mathcal{V} \int \frac{d^3p}{(2\pi)^3} \frac{E_{\mathbf{p}} - E_{\mathbf{p}} e^{-\beta E_{\mathbf{p}}} + 2E_{\mathbf{p}} e^{-\beta E_{\mathbf{p}}}}{2(1 - e^{-\beta E_{\mathbf{p}}})} \quad (7)$$

$$= \frac{\mathcal{V}}{2} \int \frac{d^3p}{(2\pi)^3} E_{\mathbf{p}} \frac{1 + e^{-\beta E_{\mathbf{p}}}}{1 - e^{-\beta E_{\mathbf{p}}}} \quad (8)$$

We're after the ground state energy, which is the energy as the temperature  $T \rightarrow 0$ . In this case  $\beta = \frac{1}{k_B T} \rightarrow \infty$  and  $e^{-\beta E_{\mathbf{p}}} \rightarrow 0$ , so we're left with

$$E_0 = \lim_{T \rightarrow 0} \frac{\mathcal{V}}{2} \int \frac{d^3p}{(2\pi)^3} E_{\mathbf{p}} \frac{1 + e^{-\beta E_{\mathbf{p}}}}{1 - e^{-\beta E_{\mathbf{p}}}} \quad (9)$$

$$= \frac{\mathcal{V}}{2} \int \frac{d^3p}{(2\pi)^3} E_{\mathbf{p}} \quad (10)$$

which agrees with 5. The energy at  $T \neq 0$  is then given relative to the ground state energy, and this can be found by subtracting 5 from 6 which gives

$$E - E_0 = \mathcal{V} \int \frac{d^3p}{(2\pi)^3} \frac{E_{\mathbf{p}} e^{-\beta E_{\mathbf{p}}}}{1 - e^{-\beta E_{\mathbf{p}}}} \quad (11)$$

$$= \mathcal{V} \int \frac{d^3p}{(2\pi)^3} \frac{E_{\mathbf{p}}}{e^{\beta E_{\mathbf{p}}} - 1} \quad (12)$$

which gives L&B's equation 25.36.

The ground state pressure  $P_0$  is given by the thermodynamic expression

$$P_0 = - \left. \frac{\partial F}{\partial \mathcal{V}} \right|_{T=0} = \frac{E_0}{\mathcal{V}} = \frac{1}{2} \int \frac{d^3p}{(2\pi)^3} E_{\mathbf{p}} \quad (13)$$

The pressure relative to the ground state pressure is then given from 3 and 13:

$$P - P_0 = -\frac{\partial F}{\partial \mathcal{V}} + \left. \frac{\partial F}{\partial \mathcal{V}} \right|_{T=0} \quad (14)$$

$$= \frac{\mathcal{V}}{\beta} \int \frac{d^3 p}{(2\pi)^3} \ln \left( 1 - e^{-\beta E_{\mathbf{p}}} \right) \quad (15)$$

which gives L&B's equation 25.37.

We can now apply this to  $\phi^4$  theory, starting with L&B's equation 25.45 for the temperature-dependent part of  $\ln Z$ :

$$\ln Z_{(1)} = -\frac{\lambda \beta \mathcal{V}}{8} \left[ \int \frac{d^3 p}{(2\pi)^3} \frac{1}{E_{\mathbf{p}}} \left( \frac{1}{e^{\beta E_{\mathbf{p}}} - 1} \right) \right]^2 \quad (16)$$

In what follows, I'll deal only with this part of  $\ln Z$ .

We get

$$F = -\frac{1}{\beta} \ln Z \quad (17)$$

$$= \frac{\lambda \mathcal{V}}{8} \left[ \int \frac{d^3 p}{(2\pi)^3} \frac{1}{E_{\mathbf{p}}} \left( \frac{1}{e^{\beta E_{\mathbf{p}}} - 1} \right) \right]^2 \quad (18)$$

The pressure is

$$P = -\frac{\partial F}{\partial \mathcal{V}} \quad (19)$$

$$= -\frac{\lambda}{8} \left[ \int \frac{d^3 p}{(2\pi)^3} \frac{1}{E_{\mathbf{p}}} \left( \frac{1}{e^{\beta E_{\mathbf{p}}} - 1} \right) \right]^2 \quad (20)$$

As the mass  $m \rightarrow 0$ ,  $E_{\mathbf{p}} = \sqrt{\mathbf{p}^2 + m^2} \rightarrow p$ , where  $p$  is the magnitude of the 3-momentum. We're therefore faced with the integral

$$I = \int \frac{d^3 p}{(2\pi)^3} \frac{1}{p} \left( \frac{1}{e^{\beta p} - 1} \right) \quad (21)$$

$$= \frac{4\pi}{(2\pi)^3} \int_0^\infty p^2 dp \frac{1}{p} \left( \frac{1}{e^{\beta p} - 1} \right) \quad (22)$$

$$= \frac{1}{2\pi^2} \int_0^\infty dp \frac{p}{e^{\beta p} - 1} \quad (23)$$

I used Maple to do the integral, although I'd guess that it can be found in tables somewhere. The result is

$$I = \frac{1}{12\beta^2} = \frac{k_B^2 T^2}{12} \quad (24)$$

From 20 we get

$$P = -\frac{\lambda k_B^4 T^4}{8 \cdot 144} \quad (25)$$

Thus the pressure has a  $T^4$  dependence for  $m \rightarrow 0$ .

As an aside, if you're interested in the integral in 23, Maple gives it as

$$\int \frac{p}{e^{\beta p} - 1} dp = -\frac{p^2}{2} + \frac{p \ln(1 - e^{\beta p})}{\beta} + \frac{\text{polylog}(2, e^{\beta p})}{\beta^2} \quad (26)$$

The polylog is a generalization of the logarithm which is defined as a series:

$$\text{polylog}(n, x) = \sum_{k=1}^{\infty} \frac{x^k}{k^n} \quad (27)$$